Physics 30

## January 2001

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## Physics 30

## Grade 12 Diploma Examination

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## January 2001

## Physics 30

## Grade 12 Diploma Examination

## Description

Time: This examination was developed to be completed in 2.5 h ; however, you may take an additional 0.5 h to complete the examination.

This is a closed-book examination consisting of

- 37 multiple-choice and 12 numericalresponse questions, of equal value, worth $70 \%$ of the examination
- 2 written-response questions, of equal value, worth a total of $30 \%$ of the examination

This examination contains sets of related questions. A set of questions may contain multiple-choice and/or numerical-response questions.

A tear-out Physics Data Sheet is included near the back of this booklet. A Periodic Table of the Elements is also provided.

Note: The perforated pages at the back of this booklet may be torn out and used for your rough work. No marks will be given for work done on the tear-out pages.

## Instructions

- You are expected to provide your own calculator. You may use any scientific calculator or a graphing calculator approved by Alberta Learning.
- You are expected to have cleared your calculator of all information that is stored in the programmable or parametric memory.
- Use only an HB pencil for the machine-scored answer sheet.
- Fill in the information required on the answer sheet and the examination booklet as directed by the presiding examiner.
- Read each question carefully.
- Consider all numbers used in the examination to be the result of a measurement or observation.
- When performing calculations, use the values of constants provided on the tear-out sheet.
Do not use the values programmed in your calculator.
- If you wish to change an answer, erase all traces of your first answer.
- Do not fold the answer sheet.
- The presiding examiner will collect your answer sheet and examination booklet and send them to Alberta Learning.
- Now turn this page and read the detailed instructions for answering machine-scored and written-response questions.

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## Multiple Choice

- Decide which of the choices best completes the statement or answers the question.
- Locate that question number on the separate answer sheet provided and fill in the circle that corresponds to your choice.


## Example

This examination is for the subject of
A. science
B. physics
C. biology
D. chemistry

Answer Sheet
(A) (C) (D)

## Numerical Response

- Record your answer on the answer sheet provided by writing it in the boxes and then filling in the corresponding circles.
- If an answer is a value between 0 and 1 (e.g., 0.25 ), then be sure to record the 0 before the decimal place.
- Enter the first digit of your answer in the left-hand box and leave any unused boxes blank.


## Examples

## Calculation Question and Solution

If a 121 N force is applied to a 77.7 kg mass at rest on a frictionless surface, the acceleration of the mass will be
$\qquad$ $\mathrm{m} / \mathrm{s}^{2}$.
(Record your three-digit answer in the numerical-response section on the answer sheet.)

$$
\begin{aligned}
& a=\frac{F}{m} \\
& a=\frac{121 \mathrm{~N}}{77.7 \mathrm{~kg}}=1.557 \mathrm{~m} / \mathrm{s}^{2}
\end{aligned}
$$



## Calculation Question and Solution

A microwave of wavelength 16 cm has a frequency, expressed in scientific notation, of $\boldsymbol{b} \times 10^{w} \mathrm{~Hz}$. The value of $\boldsymbol{b}$ is $\qquad$ -
(Record your two-digit answer in the numerical-response section on the answer sheet.)

$$
\begin{aligned}
& f=\frac{c}{\lambda} \\
& f=\frac{3.00 \times 10^{8} \mathrm{~m} / \mathrm{s}}{0.16 \mathrm{~m}}=1.875 \times 10^{9} \mathrm{~Hz}
\end{aligned}
$$



## Written Response

- Write your answers in the examination booklet as neatly as possible.
- For full marks, your answers must address all aspects of the question.
- Descriptions and/or explanations of concepts must be correct and include pertinent ideas, diagrams, calculations, and formulas. Use formulas as they appear on the equation sheet included with this examination.
- Your answers must be presented in a well-organized manner using complete sentences, correct units, and significant digits where appropriate.
- Relevant scientific, technological, and/or societal concepts and examples must be identified and made explicit.


## Scientific Notation Question and Solution

The charge on an electron is $-\boldsymbol{a} . \boldsymbol{b} \times 10^{-c d} \mathrm{C}$. The values of $\boldsymbol{a}, \boldsymbol{b}, \boldsymbol{c}$, and $\boldsymbol{d}$ are $\qquad$ , __,
$\qquad$ and $\qquad$ -
(Record all four digits of your answer in the numerical-response section on the answer sheet.)
Answer: $q=-1.6 \times 10^{-19} \mathrm{C}$


1. Which of the following quantities is a scalar quantity?
A. Force
B. Power
C. Impulse
D. Momentum

## Numerical Response

1. A golf ball has a mass of 45.0 g . A golf club is in contact with the golf ball for $3.00 \times 10^{-4} \mathrm{~s}$, and the ball leaves the club with a speed of $72.0 \mathrm{~m} / \mathrm{s}$. The average force exerted by the club on the ball, expressed in scientific notation, is $\boldsymbol{b} \times 10^{w} \mathrm{~N}$. The value of $\boldsymbol{b}$ is $\qquad$ .
(Record your three-digit answer in the numerical-response section on the answer sheet.)

## Numerical Response

2. In a vehicle safety test, a 1580 kg truck travelling at $60.0 \mathrm{~km} / \mathrm{h}$ collides with a concrete barrier and comes to a complete stop in 0.120 s . The magnitude of the change in the momentum of the truck, expressed in scientific notation, is $\boldsymbol{b} \times 10^{w} \mathrm{~kg} \cdot \mathrm{~m} / \mathrm{s}$. The value of $\boldsymbol{b}$ is $\qquad$ .
(Record your three-digit answer in the numerical-response section on the answer sheet.)

Use the following information to answer the next question.

At a coal mine, a train engine bumps an empty hopper car that has a mass of $1.00 \times 10^{4} \mathrm{~kg}$ such that it rolls at a constant speed of $2.00 \mathrm{~m} / \mathrm{s}$ under a coal storage bin. When the hopper car triggers an electromagnetic switch on the track below the storage bin, the bin drops a load of $1.20 \times 10^{4} \mathrm{~kg}$ of coal into the hopper car.


## Numerical Response

3. The speed of the hopper car immediately after receiving the load of coal, expressed in scientific notation, is $\boldsymbol{b} \times 10^{-w} \mathrm{~m} / \mathrm{s}$. The value of $\boldsymbol{b}$ is $\qquad$ .
(Record your three-digit answer in the numerical-response section on the answer sheet.)
4. The electric field between a positive point charge and a negative point charge is represented by
A.

B.

C.


D.


Use the following information to answer the next three questions.

Diagrams 1 and 2 below each show an electron as it enters a field. The fields are different but the electrons enter them with the same instantaneous velocity, $\vec{v}_{\text {inst }}$.

## Diagram 1



## Diagram 2



Statements About the Motion of the Charged Particles as They Travel Through the Fields

I The speed of the particle remains constant.
II The speed of the particle increases.
III The direction of the particle's motion remains constant.
IV The direction of the particle's motion changes.
3. The statements that describe the motion of the charged particle in diagram $\mathbf{1}$ are
A. I and III
B. I and IV
C. II and III
D. II and IV
4. The statements that describe the motion of the charged particle in diagram $\mathbf{2}$ are
A. I and III
B. I and IV
C. II and III
D. II and IV
5. The direction of the uniform magnetic field in diagram $\mathbf{2}$ is
A. toward the top of the page
B. toward the bottom of the page
C. to the left of the page
D. to the right of the page

Use the following information to answer the next four questions.

A student is given a circuit and a voltmeter. A schematic diagram of the circuit is shown below.


With the switch closed, the student records the following observations.

$$
\begin{array}{ll}
\text { Ammeter readings } & \mathrm{A}_{\mathrm{C}}=2.73 \mathrm{~mA} \\
& \mathrm{~A}_{\mathrm{E}}=1.64 \mathrm{~mA}
\end{array}
$$

Voltmeter readings between $\quad \mathbf{A}$ and $\mathbf{B}=6.00 \mathrm{~V}$

$$
\mathbf{C} \text { and } \mathbf{H}=3.27 \mathrm{~V}
$$

6. The student connects the voltmeter to the circuit at two points. A connection that produces a reading other than 3.27 V is at
A. points $\mathbf{D}$ and $\mathbf{E}$
B. points $\mathbf{D}$ and $\mathbf{H}$
C. points $\mathbf{F}$ and $\mathbf{G}$
D. points $\mathbf{G}$ and $\mathbf{H}$
7. The current through point $\mathbf{F}$ is
A. $\quad 1.09 \mathrm{~mA}$
B. $\quad 1.64 \mathrm{~mA}$
C. $\quad 2.73 \mathrm{~mA}$
D. $\quad 4.36 \mathrm{~mA}$
8. The value of the unknown resistor $R_{1}$ is
A. $\quad 1.20 \times 10^{3} \Omega$
B. $\quad 1.99 \times 10^{3} \Omega$
C. $\quad 3.00 \times 10^{3} \Omega$
D. $\quad 5.50 \times 10^{3} \Omega$
9. The total resistance of the circuit is
A. $\quad 5.45 \times 10^{2} \Omega$
B. $8.33 \times 10^{2} \Omega$
C. $2.20 \times 10^{3} \Omega$
D. $5.99 \times 10^{3} \Omega$

## Numerical Response

4. Two charged objects experience a force of 18.0 N when they are placed $5.00 \times 10^{-2} \mathrm{~m}$ apart. If the charge on one object is $1.30 \times 10^{-5} \mathrm{C}$, then the charge on the other object is $\boldsymbol{a} . \boldsymbol{b} \boldsymbol{c} \times 10^{-\boldsymbol{d}} \mathrm{C}$. The values of $\boldsymbol{a}, \boldsymbol{b}, \boldsymbol{c}$, and $\boldsymbol{d}$ are $\qquad$ , $\qquad$ , $\qquad$ , and $\qquad$ _.
(Record all four digits of your answer in the numerical-response section on the answer sheet.)

Use the following information to answer the next question.

Three types of radiation pass through an electric field along the paths shown below.

10. The types of radiation taking paths $\mathbf{X}, \mathbf{Y}$, and $\mathbf{Z}$ are, respectively,
A. beta, alpha, and gamma
B. beta, gamma, and alpha
C. gamma, alpha, and beta
D. alpha, gamma, and beta

Use the following information to answer the next question.

## An Experiment to Determine the Resistance of a Metal Wire



The points plotted on the graph above represent the results obtained from an experiment performed by a student.
11. The best completed graph of this data is
A.

B.

C.

D.


Use the following information to answer the next seven questions.

The Deep Space 1 mission (DS1) uses a ion propulsion system (IPS) on the DS1 capsule. The IPS involves ionizing atoms of xenon, accelerating them through an electric field produced by electrified grids, and ejecting the ions into space behind the capsule.

## IPS Chamber of the DS1 Capsule



In the IPS chamber, high-speed electrons collide with xenon atoms. These collisions can ionize xenon atoms. The electric field then accelerates the ions and ejects them from the IPS chamber, which propels the DS1 capsule forward.

IPS Operating Specifications for DS1
propellant ions
total mass of propellant
$\mathrm{Xe}^{+}$
mass of DS1 capsule (without propellant)
81.5 kg
energy required to ionize a xenon atom 489.5 kg
mass of a single xenon atom
exit speed of xenon ions
12.1 eV
$2.18 \times 10^{-25} \mathrm{~kg}$
$43.0 \mathrm{~km} / \mathrm{s}$
12. The minimum electron speed necessary to ionize xenon atoms is
A. $2.66 \times 10^{31} \mathrm{~m} / \mathrm{s}$
B. $\quad 5.15 \times 10^{15} \mathrm{~m} / \mathrm{s}$
C. $4.25 \times 10^{12} \mathrm{~m} / \mathrm{s}$
D. $2.06 \times 10^{6} \mathrm{~m} / \mathrm{s}$
13. The electric potential difference across the electrified grids that is required to accelerate a xenon ion from rest to its exit speed is
A. $2.93 \times 10^{-5} \mathrm{~V}$
B. $1.26 \times 10^{-3} \mathrm{~V}$
C. $1.26 \times 10^{3} \mathrm{~V}$
D. $\quad 4.71 \times 10^{29} \mathrm{~V}$
14. If all of the xenon propellant could be expelled in a single short burst, the change in the speed of the DS1 capsule after all the fuel has been exhausted would be
A. $\quad 6.14 \mathrm{~m} / \mathrm{s}$
B. $\quad 7.16 \mathrm{~m} / \mathrm{s}$
C. $\quad 6.14 \times 10^{3} \mathrm{~m} / \mathrm{s}$
D. $7.16 \times 10^{3} \mathrm{~m} / \mathrm{s}$
15. The physics principle that best describes the propulsion of the DS1 capsule is the Law of Conservation of
A. Charge
B. Energy
C. Momentum
D. Nucleon Number

## Numerical Response

5. As xenon ions in the exhaust stream behind the DS1 capsule interact with other charged particles in space, the xenon ions become neutral atoms, and in the process, emit photons. The maximum frequency of these photons, expressed in scientific notation, is $\boldsymbol{b} \times 10^{w} \mathrm{~Hz}$. The value of $\boldsymbol{b}$ is $\qquad$ .
(Record your three-digit answer in the numerical-response section on the answer sheet.)

Use the following additional information to answer the next two questions.

One isotope of xenon, xenon-133, is an unstable isotope that undergoes beta decay and has a half-life of 5.24 days.

## Numerical Response

6. If the IPS uses 81.5 kg of xenon-133 as a propellant and the launch is delayed by 26.2 days, the amount of xenon- 133 that would remain is $\qquad$ kg.
(Record your three-digit answer in the numerical-response section on the answer sheet.)
7. The decay equation for xenon-133 is
A. $\quad{ }_{54}^{133} \mathrm{Xe} \rightarrow{ }_{54}^{133} \mathrm{Xe}+\gamma$
B. $\quad{ }_{54}^{133} \mathrm{Xe} \rightarrow{ }_{52}^{129} \mathrm{Te}+{ }_{2}^{4} \alpha$
C. $\quad{ }_{54}^{133} \mathrm{Xe} \rightarrow{ }_{55}^{133} \mathrm{Cs}+{ }_{-1}^{0} \beta$
D. $\quad{ }_{54}^{133} \mathrm{Xe} \rightarrow{ }_{53}^{133} \mathrm{I}+{ }_{-1}^{0} \beta$

Use the following information to answer the next question.

17. The magnetic field induced around the rubber rod as it moves is represented by
A.

B.

C.

D.


## Numerical Response

7. An alpha particle travels at $1.08 \times 10^{5} \mathrm{~m} / \mathrm{s}$ perpendicularly through a magnetic field of strength $1.12 \times 10^{-3} \mathrm{~T}$. The magnitude of the magnetic force on the alpha particle is $\boldsymbol{b} \times 10^{-w} \mathrm{~N}$. The value of $\boldsymbol{b}$ is $\qquad$ .
(Record your three-digit answer in the numerical-response section on the answer sheet.)
8. A copper wire is connected to a battery so that it has a current in it. A segment of the wire is perpendicular to a horizontal 1.5 T magnetic field. The length of the wire in the magnetic field is 3.0 cm , and the mass of the wire affected by the magnetic field is 20 g . In order to suspend the segment of wire, the minimum current in the wire must be
A. $\quad 0.044 \mathrm{~A}$
B. $\quad 0.23 \mathrm{~A}$
C. $\quad 4.4 \mathrm{~A}$
D. 44 A

Use the following information to answer the next three questions.

The AC adapter for a pocket calculator contains a transformer that converts 120 volts into 3.0 volts. The pocket calculator draws 450 mA of current from the transformer. Assume that the transformer is an ideal transformer.
19. If the transformer's secondary coil has exactly 50 turns, then the number of turns in the primary coil is
A. 7 turns
B. 40 turns
C. 50 turns
D. $2.0 \times 10^{3}$ turns
20. The current in the primary coil of the adapter is
A. $\quad 0.80 \mathrm{~mA}$
B. $\quad 1.3 \mathrm{~mA}$
C. 11 mA
D. 18 A

## Numerical Response

8. The power supplied by the primary coil is $\qquad$ W.
(Record your two-digit answer in the numerical-response section on the answer sheet.)

## Numerical Response

9. A microwave signal that has a wavelength of $6.25 \times 10^{-3} \mathrm{~m}$ is created by an oscillating current in a microwave generator. The period of this microwave, expressed in scientific notation, is $\boldsymbol{b} \times 10^{-w}$ s. The value of $\boldsymbol{b}$ is $\qquad$ .
(Record your three-digit answer in the numerical-response section on the answer sheet.)

Use the following information to answer the next question.

A student holds a compass near the top of a filing cabinet and observes the direction that the needle points. When the student holds the compass near the bottom of the filing cabinet, the student observes that the compass needle is deflected $180^{\circ}$ from its direction at the top of the cabinet.
21. A possible explanation for the deflection of the compass needle is that the
A. bottom of the filing cabinet is positively charged
B. bottom of the filing cabinet is negatively charged
C. induced magnetic polarity of the bottom of the filing cabinet is opposite to that at the top of the filing cabinet
D. bottom of the filing cabinet is closer to Earth so it is more strongly magnetized than the top of the filing cabinet

## Numerical Response

10. An ultraviolet source emits electromagnetic waves with a frequency of $2.47 \times 10^{15} \mathrm{~Hz}$. Its wavelength, expressed in scientific notation, is $\boldsymbol{b} \times 10^{-w} \mathrm{~m}$. The value of $\boldsymbol{b}$ is $\qquad$ _.
(Record your three-digit answer in the numerical-response section on the answer sheet.)

Use the following information to answer the next two questions.

The different colours seen in exploding fireworks are produced using different elements.

## Predominant

 ColourRed
Green
Blue-Green
Yellow-Orange
22. Given the information above, the element that emits the lowest energy photon of visible light is
A. strontium
B. barium
C. copper
D. sodium
23. The colours are emitted by electrons that are
A. undergoing transitions to higher energy levels
B. undergoing transitions to lower energy levels
C. oscillating between energy levels
D. emitted by the nucleus
24. The energy gained by a proton that moves through a potential difference of 1.0 V is
A. $\quad 1.0 \mathrm{~J}$
B. $\quad 1.0 \mathrm{eV}$
C. $6.3 \times 10^{18} \mathrm{~J}$
D. $1.6 \times 10^{-19} \mathrm{eV}$

## Numerical Response

11. A metal has a work function of $2.91 \times 10^{-19} \mathrm{~J}$. Light with a frequency of $8.26 \times 10^{14} \mathrm{~Hz}$ is incident on the metal. The stopping voltage is $\qquad$ V.
(Record your three-digit answer in the numerical-response section on the answer sheet.)
12. If a light with a wavelength of $3.25 \times 10^{-8} \mathrm{~m}$ illuminates a metal surface with a work function of $5.60 \times 10^{-19} \mathrm{~J}$, the maximum kinetic energy of the emitted photoelectrons is
A. $\quad 5.60 \times 10^{-19} \mathrm{~J}$
B. $5.56 \times 10^{-18} \mathrm{~J}$
C. $\quad 6.12 \times 10^{-18} \mathrm{~J}$
D. $6.68 \times 10^{-18} \mathrm{~J}$

Use the following information to answer the next three questions.

A graph of data obtained from a photoelectric effect experiment is shown below.

Stopping Voltage as a Function of the Frequency of Incident Light on a Cesium Plate


Point $\mathbf{P}$ corresponds to a trial using light at the frequency indicated.
26. The type of light indicated by point $\mathbf{P}$ is
A. visible
B. infrared
C. microwave
D. ultraviolet
27. The energy of a photon of light indicated by point $\mathbf{P}$ is
A. $\quad 4.1 \mathrm{eV}$
B. $\quad 2.3 \mathrm{eV}$
C. $\quad 1.7 \mathrm{eV}$
D. $\quad 0.0 \mathrm{eV}$
28. Photons of light, as indicated by point $\mathbf{P}$, bombard the cesium plant. The maximum kinetic energy of an emitted electron is
A. $\quad 4.1 \mathrm{eV}$
B. 2.3 eV
C. $\quad 1.7 \mathrm{eV}$
D. 0.0 eV
29. The Compton experiment was significant in that it demonstrated that photons have
A. mass
B. momentum
C. wave properties
D. a speed of $3.00 \times 10^{8} \mathrm{~m} / \mathrm{s}$
30. An experiment starts with 1.45 kg of iodine-131. After 32.2 days, 90.6 g are left. The half-life of iodine-131 is
A. 32.2 days
B. $\quad 16.1$ days
C. 8.05 days
D. 4.04 days

Use the following information to answer the next four questions.

## Fusion Research

Interest in nuclear fusion is growing because of the amount of energy available from nuclear reactions. A major difficulty in producing a nuclear fusion reaction is that in order for nuclei to fuse, the nuclei must possess a large amount of kinetic energy. Under most circumstances, 0.25 MeV per nucleus is sufficient. At such high energies, the nuclear fuel is called a plasma.

The average kinetic energy of a nucleus within a plasma can be found using

$$
E_{\mathrm{k}}=\frac{3}{2} b T
$$

where $T$ is the temperature of the plasma, in Kelvin, and $b$ is a physical constant equal to $1.4 \times 10^{-23} \mathrm{~J} / \mathrm{K}$.

One method of obtaining the temperatures necessary for fusion is to use a high-intensity laser to heat a small cluster of nuclei. One such laser emits a $1.0 \times 10^{15} \mathrm{~W}$ pulse of ultraviolet radiation that lasts for $1.0 \times 10^{-12} \mathrm{~s}$. The wavelength of this laser is 280 nm .

## A Fusion Reaction Equation

$$
{ }_{1}^{2} \mathrm{H}+{ }_{1}^{3} \mathrm{H} \rightarrow \underline{\boldsymbol{X}}+\text { neutron }
$$

31. The missing product, $\boldsymbol{X}$, in the fusion reaction given above is
A. $\quad{ }_{2}^{5} \mathrm{He}$
B. ${ }_{2}^{4} \mathrm{He}$
C. ${ }_{1}^{4} \mathrm{H}$
D. $\quad{ }_{2}^{3} \mathrm{He}$
32. The main reason that the nuclei need to have such large kinetic energies is that
A. fusion releases large amounts of energy
B. fission must occur before fusion can occur
C. this kinetic energy is converted into nuclear energy
D. the nuclei must overcome a strong electrostatic repulsion
33. When the average kinetic energy of the nuclei in a plasma is 0.25 MeV , then the temperature is
A. $\quad 1.9 \times 10^{9} \mathrm{~K}$
B. $\quad 2.9 \times 10^{9} \mathrm{~K}$
C. $\quad 4.3 \times 10^{9} \mathrm{~K}$
D. $\quad 1.2 \times 10^{28} \mathrm{~K}$
34. The energy of a single photon of the ultraviolet laser is
A. $\quad 7.1 \times 10^{-19} \mathrm{~J}$
B. $\quad 1.0 \times 10^{-27} \mathrm{~J}$
C. $\quad 7.1 \times 10^{-28} \mathrm{~J}$
D. $1.9 \times 10^{-40} \mathrm{~J}$
35. The absorption spectrum of hydrogen is produced when electrons
A. emit radio frequency photons
B. emit short wavelength photons
C. jump from a higher orbital to a lower orbital
D. jump from a lower orbital to a higher orbital
36. An accelerated electron with 8.77 eV of energy strikes a mercury atom and leaves the collision with 2.10 eV of energy. The maximum frequency of light that can be emitted by the mercury atom is
A. $\quad 1.01 \times 10^{14} \mathrm{~Hz}$
B. $5.07 \times 10^{14} \mathrm{~Hz}$
C. $1.61 \times 10^{15} \mathrm{~Hz}$
D. $2.12 \times 10^{15} \mathrm{~Hz}$
37. For a hydrogen atom, the difference in radii between the sixth Bohr orbital and the second Bohr orbital is
A. $\quad 1.69 \times 10^{-9} \mathrm{~m}$
B. $\quad 8.46 \times 10^{-10} \mathrm{~m}$
C. $\quad 1.17 \times 10^{-11} \mathrm{~m}$
D. $1.32 \times 10^{-11} \mathrm{~m}$

## Numerical Response

12. An electron in a hydrogen atom is in the fourth orbital and jumps down to the second orbital. The energy released is $\qquad$ eV .
(Record your three-digit answer in the numerical-response section on the answer sheet.)

Use the following information to answer the next question.

In a physics demonstration, a student inflates a balloon by blowing into it. The end of the balloon is then tied. The balloon is rubbed with fur and develops an electrostatic charge. The balloon is placed against the ceiling and released. It remains "stuck" to the ceiling.

The teacher then presents the following challenges to the students:

- explain how the balloon received the electrostatic charge
- explain why the balloon is attracted to the ceiling
- provide a procedure that would determine if the charge on the balloon is positive or negative. Include a list of any additional equipment needed.
- provide a procedure that could be used to determine if there is a relationship between the amount of rubbing and the amount of charge developed on an inflated balloon. Include a list of any additional equipment needed.


## Written Response-15\%

1. Using concepts from Physics 30, provide a response to each of the teacher's challenges.

Marks will be awarded for the physics used to solve this problem and for the
effective communication of your response.

Use the following information to answer the next question.

## Falling Magnet Experiment

Two hollow tubes, one made of glass and the other made of aluminum, are positioned vertically. A student holds identical cylindrical magnets against the outside of the tubes and observe that neither tube attracts a magnet. Based on this observation, the student predicts that each magnet will fall through its respective tube with an acceleration of $9.81 \mathrm{~m} / \mathrm{s}^{2}$. The student and his lab partner then drop the magnets into the tubes from rest at position A , as shown below.


The students make the following observations:
The magnets do not touch the sides of the tubes as they fall. The time for the magnet to fall through the aluminum tube is much greater than is the time for the identical magnet to fall through the glass tube.

|  | Glass Tube | Aluminum Tube |
| :--- | :---: | :---: |
| Mass of magnet $(\mathrm{kg})$ | 0.150 | 0.150 |
| Tube length $(\mathrm{m})$ | 0.95 | 0.95 |
| Time for magnet to fall <br> from position A to <br> position B (s) | 0.44 | 0.76 |

## Written Response-15\%

2. Analyze the students' observations from the falling magnet experiment by

- completing the chart below. Include calculations to support the values you write in the chart
- explaining the results of this experiment in terms of Lenz's Law

Clearly communicate your understanding of the physics principles that you are using to solve this question. You may communicate this understanding mathematically, graphically, and/or with written statements.

|  | Glass <br> Tube | Aluminum <br> Tube |
| :--- | :--- | :--- |
| Potential Energy of <br> the magnet at <br> position A (J) |  |  |
| Acceleration of the <br> magnet through the <br> tube (m/s |  |  |
| Kinetic Energy of <br> the magnet at <br> position B (J) |  |  |
| Mechanical Energy <br> of the magnet at <br> position A (J) |  |  |
| Mechanical Energy <br> of the magnet at <br> position B (J) |  |  |
| Resisting Force on <br> the magnet (N) |  |  |

You have now completed the examination. If you have time, you may wish to check your answers.
CONSTANTS
Fold and tear along perforation.

## PHYSICS DATA SHEET








$$
\begin{aligned}
& F_{\mathrm{g}}=\frac{G m_{1} m_{2}}{r^{2}} \\
& g=\frac{G m_{1}}{r^{2}} \\
& F_{\mathrm{c}}=\frac{m v^{2}}{r} \\
& F_{\mathrm{c}}=\frac{4 \pi^{2} m r}{T^{2}}
\end{aligned}
$$


Momentum and Energy
$P=\frac{W}{t}=\frac{\Delta E}{t}$
Fold and tear along perforation.
Periodic Table of the Elements

| 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| IA | IIA | IIIB | IVB | vB | VIB | VIIB |  | VIIIB | VIIIB | 18 | IIB | IIIA | VA | VA | VIA | VIIA | VIIIA oro |
| $\begin{array}{\|ll\|} \hline 1 & \mathrm{H} \\ \text { 1.01 } & \\ \text { hydrogen } \end{array}$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  | $\begin{aligned} & 2 \mathrm{He} \\ & 4.00 \\ & \text { helium } \end{aligned}$ |
| $\begin{array}{\|ll\|} \hline 3 & \mathrm{Li} \\ \begin{array}{ll} 6.94 & \\ \text { lithium } \end{array} & \end{array}$ | $\begin{array}{\|ll\|} \hline 4 & \text { Be } \\ 9.01 & \\ \text { beryllium } \end{array}$ |  |  |  |  |  |  |  |  | $-\int_{-6.94}^{3} \mathrm{Li}$ | _Symbol | 5 B <br> 10.81  <br> boron  <br> 1  | $\begin{array}{\|ll} \hline 6 & \mathrm{C} \\ \left.\begin{array}{ll} 12.01 & \\ \text { carbon } & \\ \hline \end{array}\right] \end{array}$ | $\begin{array}{\|ll\|} \hline 7 & \mathrm{~N} \\ \hline \text { 14.01 } & \\ \text { nitrogen } \end{array}$ | $\begin{array}{\|ll\|} \hline 8 & \mathrm{O} \\ 16.00 & \\ \text { oxygen } & \end{array}$ | $\begin{array}{ll} \hline 9 & F \\ \begin{array}{ll} 19.00 \\ \text { fluorine } \end{array} & \end{array}$ | 10 Ne <br> 20.17 <br> neon |
| 11 Na <br> 22.99 <br> sodium | $\begin{array}{\|l\|} \hline 12 \mathrm{Mg} \\ 24.31 \\ \text { magnesium } \end{array}$ |  |  |  |  |  |  |  | Name - <br> () |  | $\begin{aligned} & \text { of the } \\ & \text { tope } \end{aligned}$ | $\begin{array}{\|ll\|} \hline 13 & \mathrm{Al} \\ 26.98 & \\ \text { aluminum } \end{array}$ | $\begin{array}{\|ll} \hline 14 & \mathrm{Si} \\ 28.09 & \\ \text { silicon } & \\ \hline \end{array}$ | $\begin{array}{\|ll\|} \hline 15 & \mathrm{P} \\ 30.97 & \\ \text { phosphorus } \end{array}$ | $\begin{array}{\|ll\|} \hline 16 & \mathrm{~S} \\ 32.06 & \\ \text { sulphur } \end{array}$ | $\begin{array}{ll} \hline 17 & \mathrm{Cl} \\ \begin{array}{l} 35.45 \\ \text { clorine } \end{array} \\ \hline \end{array}$ | $\begin{array}{\|ll\|} \hline 18 & \mathrm{Ar} \\ 39.95 & \\ \text { argon } \end{array}$ |
| $\begin{array}{\|lr\|} \hline 19 & \mathrm{~K} \\ 39.10 & \\ \text { potassium } \end{array}$ | 20 Ca <br> 40.08 <br> calcium | $\begin{array}{ll} \hline 21 & \text { SC } \\ 44.96 \\ \text { scandium } \end{array}$ | $\begin{array}{\|lr} \hline 22 & \mathrm{Ti} \\ \text { 47.90 } & \\ \text { titanium } \end{array}$ | $\begin{array}{\|ll\|} \hline 23 & \mathrm{~V} \\ 50.94 & \\ \text { vanadium } \end{array}$ | 24 Cr <br> 52.00 <br> chromium | $\begin{aligned} & 25 \mathrm{Mn} \\ & 54.94 \\ & \text { manganese } \end{aligned}$ | $\begin{array}{ll} 26 & \mathrm{Fe} \\ 55.85 \\ \text { iron } & \\ \hline \end{array}$ | $\begin{aligned} & \hline 27 \mathrm{Co} \\ & 58.93 \\ & \text { cobalt } \\ & \hline \end{aligned}$ | $\begin{array}{ll} \hline 28 & \mathrm{Ni} \\ 58.71 & \\ \text { nickel } \end{array}$ | $\begin{array}{\|l\|} \hline 29 \mathrm{Cu} \\ \begin{array}{l} 63.55 \\ \text { copper } \end{array} \\ \hline \end{array}$ | $\begin{aligned} & 30 \mathrm{Zn} \\ & 65.38 \\ & \text { zinc } \end{aligned}$ | $\begin{array}{\|l\|} \hline 31 \mathrm{Ga} \\ \begin{array}{l} 69.72 \\ \text { gallium } \end{array} \\ \hline \end{array}$ | $\begin{aligned} & 32 \mathrm{Ge} \\ & \begin{array}{l} 72.59 \\ \text { germanium } \end{array} \\ & \hline \end{aligned}$ | $\begin{array}{\|l\|} \hline 33 \mathrm{As} \\ 74.92 \\ \text { arsenic } \end{array}$ | $\begin{aligned} & 34 \mathrm{Se} \\ & 78.96 \\ & \text { selenium } \end{aligned}$ | $\begin{array}{ll} \hline 35 & \mathrm{Br} \\ 79.90 & \\ \text { bromine } \end{array}$ | $\begin{array}{\|lr\|} \hline 36 & \mathrm{Kr} \\ \hline 83.80 & \\ \text { krypton } \end{array}$ |
| 37 Rb <br> 85.47 <br> rubidium | $\begin{array}{\|ll\|} \hline 38 & \mathrm{Sr} \\ 87.62 & \\ \text { strontium } \end{array}$ | $\begin{array}{\|ll\|} \hline 39 & \mathrm{Y} \\ 88.91 & \\ \text { yttrium } \end{array}$ | $\begin{array}{\|ll\|} \hline 40 & \mathrm{Zr} \\ 91.22 & \\ \text { zirconium } \end{array}$ | 41 Nb <br> 92.91 <br> niobium | $\begin{array}{\|c\|} \hline 42 \mathrm{MO} \\ 95.94 \\ \text { molybdenum } \end{array}$ | $\begin{aligned} & \hline 43 \mathrm{TC} \\ & \text { (98.91) } \\ & \text { technetium } \end{aligned}$ | 44 Ru <br> 101.07 <br> ruthenium | $\begin{aligned} & \hline 45 \mathrm{Rh} \\ & \begin{array}{l} 102.91 \\ \text { rhodium } \end{array} \\ & \hline \end{aligned}$ | $\begin{aligned} & 46 \mathrm{Pd} \\ & 106.40 \\ & \text { palladium } \end{aligned}$ | $\begin{array}{\|l\|} \hline 47 \mathrm{Ag} \\ 107.87 \\ \text { silver } \end{array}$ | $\begin{aligned} & 48 \mathrm{Cd} \\ & 112.41 \\ & \text { cadmium } \end{aligned}$ | $\begin{array}{\|l\|} \hline 49 \quad \text { In } \\ \hline \begin{array}{l} \text { 114.82 } \end{array} \\ \text { indium } \end{array}$ | $\begin{array}{\|ll\|} \hline 50 & \mathrm{Sn} \\ 118.69 \\ \text { tin } & \\ \hline \end{array}$ | $\begin{array}{\|ll} \hline 51 & \mathrm{Sb} \\ \begin{array}{l} 121.75 \\ \text { antimony } \end{array} \end{array}$ | $\begin{array}{ll} \hline 52 & \mathrm{Te} \\ \begin{array}{l} 127.60 \\ \text { tellurium } \end{array} \\ \hline \end{array}$ | 53 I <br> 126.90 <br> iodine  | $\begin{array}{\|l\|} \hline 54 \mathrm{Xe} \\ 131.30 \\ \text { xenon } \end{array}$ |
| $\begin{array}{ll} \hline 55 & \mathrm{Cs} \\ 132.91 \\ \text { cesium } \end{array}$ | $\begin{aligned} & 56 \mathrm{Ba} \\ & 137.33 \\ & \text { barium } \end{aligned}$ | 57-71 | $\begin{aligned} & 72 \quad \mathrm{Hf} \\ & 178.49 \\ & \text { hafnium } \end{aligned}$ | 73 Ta <br> 180.95 <br> tantalum | $\begin{aligned} & \hline 74 \mathrm{~W} \\ & 183.85 \\ & \text { tungsten } \end{aligned}$ | 75 Re <br> 186.21 <br> rhenium | $\begin{aligned} & \hline 76 \mathrm{Os} \\ & 190.20 \\ & \text { osmium } \end{aligned}$ | 77 Ir <br> 192.22  <br> irdium  | $\begin{array}{ll} \hline 78 & \mathrm{Pt} \\ \begin{array}{l} 195.09 \\ \text { platinum } \end{array} \end{array}$ | $\begin{array}{\|l\|} \hline 79 \quad \mathrm{Au} \\ 196.97 \\ \text { gold } \end{array}$ | $\begin{aligned} & 80 \mathrm{Hg} \\ & 200.59 \\ & \text { mercury } \end{aligned}$ | $81 \quad \mathrm{Tl}$ 204.37 thallium | $\begin{array}{\|l\|} \hline 82 \mathrm{~Pb} \\ 207.19 \\ \text { lead } \end{array}$ | $\left.\begin{array}{\|ll\|} \hline 83 & \mathrm{Bi} \\ 208.98 \\ \text { bismuth } \end{array} \right\rvert\,$ | 84 Po <br> (208.98) <br> polonium | 85 At <br> $(209.98)$  <br> astatine  | 86 Rn <br> (222.02) <br> radon |
| $\begin{array}{\|lr\|} \hline 87 & \mathrm{Fr} \\ \begin{array}{l} \text { (223.02) } \\ \text { francium } \end{array} \\ \hline \end{array}$ | 88 Ra <br> (226.03) <br> radium | 89-103 | 104 Unq <br> (266.11) <br> unnilquadium | 105 Unp <br> (262.11) <br> unnilpentium | $106 \text { Unh }$ <br> (263.12) <br> unnilhexium | 107 Uns <br> (262.12) <br> unnilseptium | $108 \text { Uno }$ <br> (265) | 109 Une <br> (266) <br> unnilennium |  |  |  |  |  |  |  |  |  |


| $\begin{aligned} & 57 \mathrm{La} \\ & \text { 138.91 } \\ & \text { lanthanum } \end{aligned}$ | $\begin{array}{\|ll\|} \hline 58 & \mathrm{Ce} \\ 140.12 \\ \text { cerium } \end{array}$ | $\begin{array}{\|ll\|} \hline 59 & \mathrm{Pr} \\ 140.91 & \\ \text { praseodymium } \end{array}$ | $\begin{array}{\|l\|} \hline 60 \mathrm{Nd} \\ 144.24 \\ \text { neodymium } \end{array}$ | 61 Pm <br> (144.91) <br> promethium | $\begin{aligned} & 62 \mathrm{Sm} \\ & 150.35 \\ & \text { samarium } \end{aligned}$ | $\begin{array}{\|ll\|} \hline 63 & \mathrm{Eu} \\ \begin{array}{l} 151.96 \\ \text { europium } \end{array} \\ \hline \end{array}$ | 64 Gd <br> 157.25 <br> gadolinium | $\begin{aligned} & \hline 65 \mathrm{~Tb} \\ & 158.93 \\ & \text { terbium } \end{aligned}$ | $\begin{aligned} & 66 \text { Dy } \\ & \text { 162.50 } \\ & \text { dysprosium } \end{aligned}$ | $\begin{aligned} & 67 \mathrm{HO} \\ & 164.93 \\ & \text { holmium } \end{aligned}$ | $\begin{array}{ll} \hline 68 & \mathrm{Er} \\ 167.26 \\ \text { erbium } \end{array}$ | 69 Tm <br> 168.93 <br> thulium | $\begin{array}{\|cc\|} \hline 70 & \mathrm{Yb} \\ 173.04 \\ \text { ytterbium } \end{array}$ | $\begin{aligned} & \hline 71 \mathrm{Lu} \\ & 174.97 \\ & \text { lutetium } \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\begin{array}{\|ll\|} \hline 89 & \text { AC } \\ (277.03) \\ \text { actinium } \end{array}$ | $\begin{array}{ll} 90 & \mathrm{Th} \\ (232.04) \\ \text { thorium } \end{array}$ | $\begin{array}{ll} 91 & \mathrm{~Pa} \\ \begin{array}{l} (231.04) \\ \text { protactinium } \end{array} \end{array}$ | $\begin{array}{ll} \hline 92 & \mathrm{U} \\ 238.03 & \\ \text { uranium } \end{array}$ | $\begin{aligned} & 93 \mathrm{~Np} \\ & \begin{array}{l} (237.05) \\ \text { neptunium } \end{array} \end{aligned}$ | $\begin{aligned} & 94 \mathrm{Pu} \\ & \begin{array}{l} (244.06) \\ \text { plutonium } \end{array} \end{aligned}$ | $\begin{array}{\|l\|} \hline 95 \mathrm{Am} \\ \begin{array}{l} (243.06) \\ \text { americium } \end{array} \\ \hline \end{array}$ | $96 \mathrm{Cm}$ <br> (247.07) <br> curium | $\begin{aligned} & \hline 97 \mathrm{BK} \\ & (247.07) \\ & \text { berkelium } \end{aligned}$ | $\begin{array}{ll} \hline 98 \quad \mathrm{Cf} \\ \begin{array}{l} (242.06) \\ \text { californium } \end{array} \end{array}$ | $\begin{aligned} & 99 \quad \text { ES } \\ & \text { (252.08) } \\ & \text { einsteinium } \end{aligned}$ | $\begin{aligned} & \text { 100Fm } \\ & (257.10) \\ & \text { fermium } \end{aligned}$ | $\begin{aligned} & 101 \mathrm{Md} \\ & (258.10) \\ & \text { mendelevium } \end{aligned}$ | $\begin{array}{\|l\|} \hline 102 \mathrm{No} \\ \begin{array}{l} (259.10) \\ \text { nobelium } \end{array} \\ \hline \end{array}$ | $\begin{aligned} & \hline 103 \mathrm{Lr} \\ & \text { (260.11) } \\ & \text { lawrencium } \end{aligned}$ |

No marks will be given for work done on this page.
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Physics 30 January 2001 Diploma Examination Multiple Choice and Numerical Response Keys

1. B
2. B
3. C
4. B
5. B
6. D
7. A
8. B
9. C
10. B
11. B
12. D
13. C
14. D
15. C
16. C
17. C
18. C
19. D

| 1. | 1.08 |
| :--- | :--- |
| 2. | 2.63 |
| 3. | 9.09 |
| 4. | 3857 |
| 5. | 2.92 |
| 6. | 2.55 |
| 7. | 3.87 |
| 8. | 1.4 |
| 9. | 2.08 |
| $\mathbf{1 0 .}$ | 1.21 |
| $\mathbf{1 1}$ | 1.60 |
| $\mathbf{1 2}$ | 2.55 or 2.56 |

## Holistic Scoring Guide Draft

Major Concepts: Charging by contact, induced charge separation, experimental design

| Score | Criteria |
| :---: | :---: |
| 5 <br> Excellent | - The student provides a complete solution covering the full scope of the question. <br> - The reader has no difficulty following the strategy or solution presented by the student. <br> - Statements made in the response are supported explicitly but may contain minor errors or have minor omissions. |
| 4 Good | - The student provides a solution to the significant parts of the question. <br> - The reader may have some difficulty following the strategy or solution presented by the student. <br> - Statements made in the response are supported, but the support may be implicit. |
| 3 <br> Satisfactory | - The student provides a solution in which he/she has made significant progress toward answering the question. <br> - The reader has difficulty following the strategy or solution presented by the student. <br> - Statements made in the response may be open to interpretation and may lack support. |
| $\begin{gathered} 2 \\ \text { Limited } \end{gathered}$ | - The student provides a solution in which he/she has made some progress toward answering the question. <br> - Statements made in the response lack support. |
| $\begin{gathered} 1 \\ \text { Poor } \end{gathered}$ | - The student provides a solution that begins to answer the question. |
| $\begin{gathered} 0 \\ \text { Insufficient } \end{gathered}$ | - The student provides a solution that is invalid for the major concepts addressed by the question. |
| NR | No response is given. |

Use the following information to answer the next question.

In a physics demonstration, a student inflates a balloon by blowing into it. The end of the balloon is then tied. The balloon is rubbed with fur and develops an electrostatic charge. The balloon is placed against the ceiling and released. It remains "stuck" to the ceiling.

The teacher then presents the following challenges to the students:

- explain how the balloon received the electrostatic charge
- explain why the balloon is attracted to the ceiling
- provide a procedure that would determine if the charge on the balloon is positive or negative. Include a list of any additional equipment needed.
- provide a procedure that could be used to determine if there is a relationship between the amount of rubbing and the amount of charge developed on an inflated balloon. Include a list of any additional equipment needed.


## Written Response-15\%

1. Using concepts from Physics 30 , provide a response to each of the teacher's challenges.

Marks will be awarded for the physics used to solve this problem and for the effective communication of your response.

A complete response should include the following content. The clarity of the response is considered in assigning a mark.

## Expected Content:

This question requires both a theoretical and an empirical understanding of electrostatics.

## Theoretical

To completely address the theoretical aspect of the question, the student needs to address the movement of charges. The response should include the following:

- A statement that the balloon receives its charge by friction or by contact.
- A statement that electrons move between the balloon and the fur. (Students do not need to specify that the balloon will become negatively charged, or identify the charge that remains on the fur after contact.)
- A description of the charge separation that occurs on the ceiling.
- An explanation of the process of inducing a charge.
- A statement that opposite charges attract.


## Empirical

Two procedures are required to completely address the empirical aspect of the question.

- A procedure is required to determine the type of charge on the balloon. The response must include instructions and equipment needed. The procedure must result in the determination of the nature of the charge.
- A procedure is required to determine if there is a relationship between the amount of rubbing and the amount of charge developed on an inflated balloon. The response must include instructions and equipment needed. Instructions must specify repeated trials, and identify both the manipulated and responding variables. (Examples of the manipulated variable include: time of rubbing, number of strokes, or amount of pressure applied. Examples of the responding variable include: the time that the balloon remains stuck to the ceiling, divergence of the leaves of an electroscope, or force of attraction of the balloon to the ceiling.)


## Scoring Guide for Anaholistic Questions

| Major Concepts: Potential energy; Acceleration and kinetic energy; Resisting force; Lenz's Law |  |
| :---: | :--- |
| Score | Criteria | \left\lvert\, \(\left.\begin{array}{l}In the response, the student <br>

- uses an appropriate method that reflects an excellent understanding of all major concepts <br>
- provides a complete description of the method used and shows a complete solution for the <br>
problem <br>

- states formulas explicitly\end{array}\right.\right\}\)| may make a minor error, omission, or inconsistency; however, this does not hinder the |
| :--- |
| understanding of the physics content |
| - draws diagrams that are appropriate, correct, and complete |$|$

## Falling Magnet Experiment

Two hollow tubes, one made of glass and the other made of aluminum, are positioned vertically. A student holds identical cylindrical magnets against the outside of the tubes and observe that neither tube attracts a magnet. Based on this observation, the student predicts that each magnet will fall through its respective tube with an acceleration of $9.81 \mathrm{~m} / \mathrm{s}^{2}$. The student and his lab partner then drop the magnets into the tubes from rest at position A, as shown below.


The students make the following observations:
The magnets do not touch the sides of the tubes as they fall. The time for the magnet to fall through the aluminum tube is much greater than is the time for the identical magnet to fall through the glass tube.

|  | Glass Tube | Aluminum Tube |
| :--- | :---: | :---: |
| Mass of magnet $(\mathrm{kg})$ | 0.150 | 0.150 |
| Tube length $(\mathrm{m})$ | 0.95 | 0.95 |
| Time for magnet to fall <br> from position A to <br> position B (s) | 0.44 | 0.76 |

## Written Response-15\%

2. Analyze the students' observations from the falling magnet experiment by

- completing the chart below. Include calculations to support the values you write in the chart
- explaining the results of this experiment in terms of Lenz's Law

Clearly communicate your understanding of the physics principles that you are using to solve this question. You may communicate this understanding mathematically, graphically, and/or with written statements.

Sample Solution:
$\left.\begin{array}{l|c|c|} & \begin{array}{c}\text { Glass } \\ \text { Tube }\end{array} & \begin{array}{c}\text { Aluminum } \\ \text { Tube }\end{array} \\ \hline \begin{array}{l}\text { Potential Energy of } \\ \text { the magnet at } \\ \text { position A (J) }\end{array} & 1.4 & 1.4 \\ \hline \begin{array}{l}\text { Acceleration of the } \\ \text { magnet through the } \\ \text { tube (m/s }\end{array}\end{array}\right)$

- Potential Energy

$$
\begin{aligned}
E_{\mathrm{p}_{\mathrm{A}}} & =m g h \\
& =(0.150 \mathrm{~kg})\left(9.81 \mathrm{~m} / \mathrm{s}^{2}\right)(0.95 \mathrm{~m}) \\
E_{\mathrm{p}_{\mathrm{A}}} & =1.398 \mathrm{~J}
\end{aligned}
$$

- Acceleration: from $d=v_{i} t+\frac{1}{2} a t^{2} \quad a=\frac{2 d}{t^{2}}$

$$
\begin{array}{ll}
a_{\text {glass }}=\frac{2(0.95 \mathrm{~m})}{(0.44 \mathrm{~s})^{2}} & a_{\text {aluminum }}=\frac{2(0.95 \mathrm{~m})}{(0.76 \mathrm{~s})^{2}} \\
a_{\text {glass }}=9.814 \mathrm{~m} / \mathrm{s}^{2} & a_{\text {aluminum }}=3.289 \mathrm{~m} / \mathrm{s}^{2}
\end{array}
$$

- Kinetic Energy

$$
\begin{array}{rlrl}
v_{\mathrm{f}}^{2} & =v_{\mathrm{i}}^{2}+2 a d & v_{\mathrm{f}}^{2}=2 a d \operatorname{since} v_{\mathrm{i}}=0 \\
E_{\mathrm{k}} & =\frac{1}{2} m v_{\mathrm{f}}^{2} & E_{\mathrm{k}}=\frac{1}{2} m\left(2 a_{\mathrm{net}} d\right)=m a_{\mathrm{net}} d \\
E_{\text {glass }} & =m a_{\mathrm{net}} d & E_{\text {aluminum }} & =m a_{\mathrm{net}} d \\
& =(0.150 \mathrm{~kg})\left(9.8 \mathrm{~m} / \mathrm{s}^{2}\right)(0.95 \mathrm{~m}) & & =(0.150 \mathrm{~kg})\left(3.3 \mathrm{~m} / \mathrm{s}^{2}\right)(0.95 \mathrm{~m}) \\
E_{\text {glass }} & =1.397 \mathrm{~J} & E_{\text {aluminum }} & =0.470 \mathrm{~J}
\end{array}
$$

Students can calculate $v_{\mathrm{f}}$ from $d=\frac{v_{\mathrm{i}}+v_{\mathrm{f}}}{2} t$. Since $v_{\mathrm{i}}=0, v_{\mathrm{f}}=\frac{2 d}{t}$. As a result:

$$
v_{\mathrm{f}(\mathrm{glass})}=4.32 \mathrm{~m} / \mathrm{s}, \text { and } v_{\mathrm{f}(\text { aluminum })}=2.50 \mathrm{~m} / \mathrm{s}
$$

- Resisting Force


## Method 1

$$
\begin{aligned}
\Delta E & =\text { Work done }=F d & & \\
F_{\text {glass }} & =\frac{\Delta E}{d} & F_{\text {aluminum }} & =\frac{\Delta E}{d} \\
& =\frac{1.4 \mathrm{~J}-1.4 \mathrm{~J}}{0.95 \mathrm{~m}} & & =\frac{1.4 \mathrm{~J}-0.47 \mathrm{~J}}{0.95 \mathrm{~m}} \\
F_{\text {glass }} & =0 \mathrm{~N} & F_{\text {aluminum }} & =0.979 \mathrm{~N}
\end{aligned}
$$

## Method 2

$$
\begin{aligned}
F_{\mathrm{net}} & =m a \\
F_{\mathrm{g}}-F_{\mathrm{A}} & =m a \\
(0.150 \mathrm{~kg})\left(9.81 \mathrm{~m} / \mathrm{s}^{2}\right)-F_{\mathrm{A}} & =(0.150 \mathrm{~kg})\left(3.29 \mathrm{~m} / \mathrm{s}^{2}\right) \\
F_{\mathrm{A}} & =0.98 \mathrm{~N}
\end{aligned}
$$

## - Lenz's Law Explanation

By Lenz's Law, a changing magnetic field will cause an induced magnetic field in the aluminum tube that opposes the original changing magnetic field.

Since the glass tube is an electrical insulator, there is no induced magnetic field.

